

which is characteristic for a *trans* orientation.

The compounds 5-8 are accessible from reaction of 3 with N-chlorosuccinimide (NCS) ⁶, N-bromosuccinimide (NBS) or cyanogenbromide, iodine and methyl iodide respectively.

A typical procedure is as follows :

To 0.010 mole of the cuprate 2, prepared as described previously ^{7,8} in 30 ml of dry THF at -50°C - -60°C , a solution of 0.009 mole of 1 in 10 ml of dry THF was added. After stirring during 15 minutes at -50°C , the temperature was slowly raised to $+20^{\circ}\text{C}$. The reaction with water, cyanogen bromide, N-chloro- or N-bromosuccinimide, iodine or methyl iodide was carried out at $+20^{\circ}\text{C}$ (in THF).

Synthesis of $\begin{array}{c} \text{Ph}_3\text{Si} \\ \diagdown \\ \text{C}=\text{C} \\ \diagup \\ \text{X} \end{array} \begin{array}{c} \text{H} \\ \diagup \\ \text{C} \\ \diagdown \\ \text{R} \end{array}$ from 1 and 2 ⁹. (Yields : 80 -95 %)

R	X	m.p. ($^{\circ}\text{C}$)
Et	H	102 - 103
i-Pr	H	84 - 85
c-Hexyl	H	52 - 53
t-Bu	H	80 - 81
Et	Br	79 - 80
Et	I	97 - 98
i-Pr	Me	46 - 48
t-Bu	Cl	85 - 86

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- The same results were obtained when a complex of 2 and LiBr was used (cf ⁸⁻⁹).